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compounds contain a central

atom or cation which is

coordinated to a suitable

number of anions or neutral

molecules and usually retain

their identity in solution as well

as in solid state these may be a

positively charged negatively

charged or a neutral

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ferromagnetic substances

substances which are attracted

very strongly by a magnetic

field are called ferromagnetic

substances e g fe ni co and cro

2 show ferromagnetism

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openstax Feb 29 2020 web remember that a number raised to the zero power is equal to 1 thus $a^0 = 1$ which is why the concentration term may be omitted from the rate law the rate of reaction is solely dependent on the concentration of NO_2 a later chapter section on reaction mechanisms will explain how a reactant's concentration can

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carboxylic acid contains a

carbonyl functional group and

hydroxyl group it interact

easily with polar compounds

and contributes to many

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describes the frequently used

components in dairy the

phenomenon of osmosis can be

illustrated by the example

shown in figure 2 12 in

chemistry such a system is

called a buffer solution because

within certain limits the pH value remains constant when acids or bases are added this effect can be explained by **mcq questions for class 12 chemistry chapter 11 alcohols phenols** Mar 31 2020 web 7 9 2020 check the below ncert mcq questions for class 12 chemistry chapter 11 alcohols phenols and ethers with answers pdf free download mcq questions for class 12 chemistry with answers were prepared based on the latest exam pattern we have provided alcohols phenols and ethers class 12 chemistry mcqs questions with *ck12 foundation* Jun 02 2020 web flexbook platform flexbook flexlet and flexcard are registered trademarks of ck 12 foundation **15 1 precipitation and dissolution chemistry 2e openstax** Aug 24 2019 web recall from the chapter on solutions that the solubility of a substance can vary from essentially zero insoluble or sparingly soluble to infinity miscible a solute with finite solubility can yield a saturated

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or a gaseous solution 3
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